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U.S. DEPARTMENT OF COMMERCE / National Bureau of Standards

A Catalog of Data Compilations on Photochemical and Photophysical Processes in Solution

78

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THE NATIONAL MEASUREMENT LABORATORY provides the national system of physical and chemical and materials measurement; coordinates the system with measurement systems of other nations and furnishes essential services leading to accurate and uniform physical and chemical measurement throughout the Nation's scientific community, industry, and commerce; conducts materials research leading to improved methods of measurement, standards, and data on the properties of materials needed by industry, commerce, educational institutions, and Government; provides advisory and research services to other Government agencies; develops, produces, and distributes Standard Reference Materials; and provides calibration services. The Laboratory consists of the following centers:

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[2]Some divisions within the center are located at Boulder, CO 80303.

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A Catalog of Data Compilations on Photochemical and Photophysical Processes in Solution

James G. Brummer, W. Phillip Helman,
and Alberta B. Ross

Radiation Chemistry Data Center
Radiation Laboratory
University of Notre Dame
Notre Dame, Indiana 46556

U.S. DEPARTMENT OF COMMERCE, Philip M. Klutznick, Secretary

Luther H. Hodges, Jr., Deputy Secretary

Jordan J. Baruch, Assistant Secretary for Productivity, Technology and Innovation

NATIONAL BUREAU OF STANDARDS, Ernest Ambler, Director

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Foreword

The National Standard Reference Data System was established in 1963 for the purpose of promoting the critical evaluation and dissemination of numerical data of the physical sciences. The program is coordinated by the Office of Standard Reference Data of the National Bureau of Standards but involves the efforts of many groups in universities, government laboratories, and private industry. The primary aim of the program is to provide compilations of critically evaluated physical and chemical property data. These tables are published in the *Journal of Physical and Chemical Reference Data,* in the NSRDS-NBS series of the National Bureau of Standards, and through other appropriate channels.

The task of critical evaluation is carried out in various data centers, each with a well-defined technical scope. A necessary preliminary step to the critical evaluation process is the retrieval from the world scientific literature of all papers falling within the scope of the center. Each center, therefore, builds up a comprehensive well-indexed bibliographical file which forms the base for the evaluation task. Bibliographies derived from these files are published when they appear to be of value to research workers and others interested in the particular technical area.

Further information on NSRDS and the publications which form the primary output of the program may be obtained by writing to the Office of Standard Reference Data, National Bureau of Standards, Washington, DC 20234.

David R. Lide, Jr., Chief
Office of Standard Reference Data

A Catalog of Data Compilations on Photochemical and Photophysical Processes in Solution

James G. Brummer, W. Phillip Helman and Alberta B. Ross
Radiation Chemistry Data Center, Radiation Laboratory, University of Notre Dame, Notre Dame, Indiana 46556*

References to compilations and reviews of data on photochemical and photophysical processes in solution have been annotated to indicate subject and data content. Indexes are included for data types, keywords and authors.

Keywords: Data compilations; photochemistry; photophysics; review; solutions.

Introduction

In 1971 a Report of the CODATA Task Group for Chemical Kinetics[1] listed 228 reviews, compilations and evaluations of kinetic data which had been published, or which were planned, in preparation or in press. That catalog included only a few publications on photochemistry in solution. Since then the Chemical Kinetics Information Center[2] and other groups sponsored by the National Standard Reference Data System[3] have carried out a number of compilations and evaluations of gas phase kinetic data relevant to photochemistry. The present survey of availability of data compilations on photochemical and photophysical processes in solution has been undertaken in order to find areas where tabulations and reviews of such data have been made; the catalog will provide a reference aid for locating collections of such data.

The Radiation Chemistry Data Center collects and indexes current literature within its scope, which has recently been expanded to include photochemical and photophysical processes in solution. The citations herein are part of the RCDC Bibliographic Data Base which contains 45,000 references (mainly since 1966) on radiation chemistry and photochemistry. A careful examination was made of our files and other sources; we hope that readers will call our attention to compilations which should have been included. Although this catalog does not include unpublished works, we should also like to know of compilations in preparation or in press.

The present catalog has been limited to those publications which contain substantial tabulations of data. Some publications here cited contain kinetic data which were obtained by methods other than photochemistry, but which may be relevant to photochemical processes. Compilations are also included on spectroscopy of excited electronic states and transient radicals and radical ions in solution. Reviews have not been cited where data were included mainly for illustration; likewise, textbooks are omitted. References to publications available only in Japanese or Russian have also been omitted. In addition to the references cited here, our files show that several hundred reviews on photochemistry have been published in a wide variety of periodicals, books and serials. There are a number of review series devoted to photochemical topics which are listed in the

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[1] A Catalog of Compilation and Data Evaluation Activities in Chemical Kinetics, Photochemistry and Radiation Chemistry, by D. Garvin, *CODATA Bulletin*, Issue No. 3, Dec. 1971, 28p.

[2] National Bureau of Standards, Washington, DC, Dr. Robert Hampson, Jr., Director

[3] See: Critical Evaluation of Data in the Physical Sciences – A Status Report on the National Standard Reference Data System, January 1977. S.A. Rossmassler, Ed., NBS Technical Note 947, May 1977, 84p.

Appendix. Only a few of the reviews in those series are cited here. For current reviews the reader is referred to the references listed in category Z of the *Biweekly List of Papers on Radiation Chemistry and Photochemistry* [4].

Arrangement of References and Indexes

References to the data compilations have been grouped by year of publication. The reference list is followed by author, data type and keyword indexes. Each entry in the reference list has been given a number for convenience in preparing the indexes; the first two digits of the number represent the year of publication. Each entry contains one or more symbols indicating the type of data included. A list of symbols for data type follows this section. Symbols and abbreviations are also given for modifiers which are used in parentheses or as subscripts to the data types to represent processes or states; thus, τ_P = phosphorescence lifetime. ϕ_{aq} = quantum yield for aquation, and ϕ_F = fluorescence quantum yield are data identifiers combining a data type with a process.

Following the data identifiers are keywords representing classes of materials, compound types, or types of intermediates. The next line of the entry contains numbers of references, tables and figures. In some cases a further annotation was added. All of the references cover liquid phase solutions; where gas, pure liquid, or aqueous systems were also covered those terms are present as keywords. Indexes have been prepared using the data types and keywords associated with each entry. An author index is also included.

List of Symbols and Abbreviations

Data Types

ϕ	quantum yield
prod. anal.	product analysis
E	energy level
ΔE	energy gap
abs. spec.	absorption spectra
emis. spec.	emission spectra
excit. spec.	excitation spectra
ϵ_{abs}	absorptivity
ϵ_{emis}	emissivity
λ_{abs}	absorption maximum
λ_{emis}	emission maximum
f	oscillator strength
τ	lifetime
$t_{1/2}$	half-life
k	rate constant
A	preexponential factor
E_a	activation energy
K	equilibrium constant
pK	negative logarithm of the acid dissociation constant
E°	redox potential
I	luminescence intensity
I.P.	ionization potential
μ	dipole moment
R	energy transfer distance

Processes and States

ab	abstraction
add	addition
	adduct formulation
	association
aq	aquation
B	biradicals
com	combination
d	decay
dim	dimerization
dis	dissociation
disprop	disproportionation
elim	elimination
et	electron transfer
ET	energy transfer
EXM	excimer
F	fluorescence
H-ab	H abstraction
hal-ab	halogen abstraction
ic	internal conversion
isc	intersystem crossing
isom	isomerization
	rearrangement
nr	nonradiative
ox	oxidation

[4] *Biweekly List of Papers on Radiation Chemistry and Photochemistry* Notre Dame, IN, Radiation Chemistry Data Center, biweekly wit annual indexed cumulation.

P	phosphorescence	red	reduction
Q	quartet states	redox	oxidation–reduction
q	quenching	sub	substitution
r	radiative	S	singlet states
rac	racemization	T	triplet states

References

64–01 Wilkinson, F. Electronic energy transfer between organic molecules in solution. Advances in Photochemistry, W.A. Noyes, G.S. Hammond and J.N. Pitts, Jr. (eds.), Interscience Publishers, New York, N.Y., 1964, Vol. 3, p.241–268

k(T–T); τ_S; ΔE(T–T)
energy transfer; quenching; aromatics; singlet states; triplet states
98 ref.; 5 tab.; 4 fig.

65–01 Berlman, I.B. Handbook of fluorescence spectra of aromatic molecules. Academic Press, New York, N.Y., 1965, 258p.

ϕ_F; τ_F; ϵ_{emis}; ϵ_{abs}; λ_{emis}; λ_{abs}; abs. spec.; emis. spec.
fluorescence; aromatics; heterocyclic compounds
~100 ref.; 147 spectra; See later edition [71–04]. Extensive bibliography containing references to 1000 works.

67–01 Adamson, A.W. Photochemistry of complex ions. IV. The role of quartet excited states in the photochemistry of chromium(III) complexes. J. Phys. Chem. 71(4): 798–808 (1967)

ϕ_{aq}; ϕ_{isom}; E_a
quartet states; transition metal complexes; chromium compounds; aqueous
26 ref.; 3 tab.; 3 fig.

67–02 Balzani, V.; Moggi, L.; Scandola, F.; Carassiti, V. Photochemistry of cobalt(III) complexes. Inorg. Chim. Acta Rev. 1: 7–34 (1967)

ϕ_{redox}; ϕ_{aq}; ϕ_{isom}; ϕ_{rac}
transition metal complexes; cobalt compounds; aqueous
111 ref.; 6 tab.; 8 fig.

67–03 Birks, J.B.; Munro, I.H. The fluorescence lifetimes of aromatic molecules. Progress in Reaction Kinetics, Porter, G. (ed.), Pergamon Press, New York, NY, 1967, Vol. 4, p.239–303

ϕ_F; k_r; k_{dis}(EXM); τ_F
solid; polycyclic; dyes; biological materials; aromatics; fluorescence; excimer
257 ref.; 10 tab.; 16 fig.

67–04 Schmillen, A.; Legler, R. Luminescence of organic substances. Landolt–Börnstein, Numerical Data and Functional Relationships in Science and Technology. New Series, Group II: Atomic and Molecular Physics, K.–H. Hellwege (ed.), Springer–Verlag, Berlin, 1967, Vol. 3, 416p.

ϕ_r; k_q; k_{ET}; τ_r; λ_{emis}; λ_{abs}; emis. spec.; μ; R
fluorescence; phosphorescence; transition metal complexes; solid; singlet–triplet transitions; singlet–singlet transitions; quenching; excimer; organic; aromatics; excited states; triplet–triplet transitions; triplet states; energy transfer; singlet states
~1000 ref.; many tables and figures; index; main table includes 920 compounds

67–05 Anbar, M.; Neta, P. A compilation of specific bimolecular rate atom; hydroxyl radical bimolecular rate constants for the reactions of hydrated electrons, hydrogen atoms and hydroxyl radicals with inorganic and organic compounds in aqueous solution. Intern. J. Appl. Radiation Isotopes 18: 493–523 (1967)

k
aqueous; hydrated electron; hydrogen atom; hydroxyl radical
164 ref.; 3 tab.

68–01 Habersbergerova, A.; Janovsky, I.; Teply, J. Absorption spectra of intermediates formed during radiolysis and photolysis. Radiat. Res. Rev. 1: 109–81 (1968)

ϵ_{abs}; λ_{abs}; $t_{1/2}$
solvated electrons; radicals; organic; inorganic; solid; heterocyclic compounds; liquid; aliphatics; aromatics
201 ref.; 12 tab.; Extensive compilation of data on radical transients (see [72–01]). Data for both aqueous and nonaqueous solution.

68–02 Valentine, D.,Jr. The photochemistry of cobalt(III) and chromium(III) complexes in solution. Advances in Photochemistry, W.A. Noyes, Jr., G.S. Hammond and J.N. Pitts, Jr. (eds.), John Wiley and Sons, New York, N.Y., 1968, Vol. 6, p.123–92

ϕ_P; ϕ_{aq}; ϕ_{rac}; ϕ_{red}; ϕ_{sub}; τ_P; E_a
solid; cobalt compounds; chromium compounds; aqueous; phosphorescence; transition metal complexes
151 ref.; 4 tab.; 7 fig.

68–03 Arnold, D.R. The photocycloaddition of carbonyl compounds to unsaturated systems: The syntheses of oxetanes. Advances in Photochemistry, W.A. Noyes, Jr., G.S. Hammond and J.N. Pitts, Jr. (eds.), John Wiley and Sons, New York, N.Y., 1968, Vol. 6, p.301–423

ϕ_{add}; E_T; prod. anal.
alkenes; carbonyl compounds; triplet states; solid; ketones; aldehydes
166 ref.; 22 tab.; 6 fig.

68–04 Adamson, A.W.; Waltz, W.L.; Zinato, E.; Watts, D.W.; Fleischauer, P.D.; Lindholm, R.D. Photochemistry of transition–metal coordination compounds. Chem. Rev. 68: 541–85 (1968)

ϕ_{rac}; ϕ_{aq}; ϕ_{redox}; λ_{abs}; ϵ_{abs}; E_a
transition metal complexes; cobalt compounds; rhodium compounds; platinum compounds; iron compounds; tungsten compounds; chromium compounds; aqueous
472 ref.; 8 tab.; 13 fig.

69–01 Dilling, W.L. Photochemical cycloaddition reactions of nonaromatic conjugated hydrocarbon dienes and polyenes. Chem. Rev. 69(6): 845–77 (1969)

ϕ_{dim}; ϕ_{add}; E_T; prod. anal.
dienes; polyenes; carbonyl compounds; ketones; triplet states; aromatics
149 ref.; 22 tab.; 3 fig.

69–02 Turro, N.J. Photochemical reactions of organic molecules. Tech. Org. Chem. 14: 133–296 (1969)

ϕ_{isc}; ϕ_{red}; k_{isc}; ΔE(S–T); E_S; E_T
singlet states; triplet states; carbonyl compounds; alkenes; aromatics; ketones; organic; dienes; solid; polyenes
722 ref.; 11 tab.; 16 fig.

69–03 McGlynn, S.P.; Azumi, T.; Kinoshita, M. Molecular spectroscopy of the triplet state. Prentice–Hall, Inc., Englewood Cliffs, N.J., 1969, 434p.

ϕ_F; ϕ_P; ϕ_{ic}; k_F; k_{isc}; τ_P; E_T; ΔE(S–T); ΔE(T–T); f
triplet states; phosphorescence; aromatics; solid; singlet states; carbonyl compounds; azo–compounds; alkenes; benzene; dyes; heterocyclic compounds; nitro compounds; transition metal complexes; fluorescence
1233 ref.; 434 page book containing numerous tables of data and figures.

69–04 Becker, R.S. Theory and interpretation of fluorescence and phosphorescence. Wiley Interscience, New.York, N.Y. 1969, 283p.

ϕ_F; ϕ_P; ϕ_{isc}; τ_F; τ_P; τ_T; ΔE(S–T); pK(S); pK(T); λ_{emis}
aromatics; polycyclic; phosphorescence; fluorescence; heterocyclic compounds; singlet states; triplet states; solid; biological materials; porphyrins; transition metal complexes
525 ref.; 27 tab.; Data for both aqueous and nonaqueous solution.

70–01 Birks, J.B. Photophysics of Aromatic Molecules. Wiley Interscience, New York. 1970, 704p.

ϕ_r; ϕ_{isc}; ϕ_F; ϕ_T; ϕ_P; k_q; k_F; k_{isc}; k_{ic}; k_P; k_{dim}(EXM); τ_T; τ_F; τ_r; E_a; E_T; ΔE(T–T); ΔE(S–T); ϵ_{abs}(T–T); ϵ_{abs}(S–S); λ_{abs}; λ_{emis}; I.P.; R
aromatics; singlet states; triplet states; excimer; fluorescence; phosphorescence; exciplex; energy transfer; quenching; gas; solid: excited states
704 page book citing ~1100 ref.. Contains numerous tables of data and many figures. Data for both aqueous and nonaqueous solution.

70–02 Dalton, J.C.; Turro, N.J. Photoreactivity of n.π^* excited states of alkyl ketones. Annu. Rev. Phys. Chem. 21: 499–560 (1970)

ϕ_{isom}; ϕ_{red}; ϕ_{elim}; ϕ(prod); k_q; τ_F; τ_T; τ_S; prod. anal.
ketones; aliphatics; fluorescence; singlet states; triplet states; quenching; Norrish type I processes; Norrish type II processes
102 ref.; 14 tab.; 9 fig.

70–03 Birks, J.B. Excimer fluorescence of aromatic compounds. Progress in Reaction Kinetics, G. Porter, K.R. Jennings and P. Suppan (eds.), Pergamon Press, New York, N.Y., 1970, Vol. 5, p.181–272

ϕ_{nr}; ϕ_F; ϕ_T; k_F; k_{dim}(EXM); k_{dis}(EXM); k_{nr}; k_{ET}; k_q; τ_{nr}; τ_F; E_a; λ_{abs}; λ_{emis}; f
aromatics; fluorescence: excimer; exciplex; liquid; solid
245 ref.; 23 tab.; 21 fig.

70–04 Vander Donckt, E. Acid–base properties of excited states. Progress in Reaction Kinetics, G. Porter, K.R. Jennings and P. Suppan (eds.), Pergamon Press, New York, N.Y.. 1970, Vol. 5, p.273–99

pK(S); pK(T)
singlet states; triplet states; aromatics; amines; phenols, substituted; carboxylic acids; acidity
98 ref.; 10 tab.; 11 fig.

70–05 Denisov, E.T. Interactions between atoms and radicals in the liquid phase. Russ. Chem. Rev. 39(1): 31–46 (1970)

k_{com}; $k_{disprop}$; k; E_a; A
radicals; alkyl radicals; liquid; alkoxy radicals; phenols, substituted; peroxy radicals; radical processes; aryloxy radicals
165 ref.; 9 tab.; Data for both aqueous and nonaqueous solution.

71–01 De Mayo, P. Enone photoannelation. Acc. Chem. Res. 4(2): 41–7 (1971)

k_d; k_{add}; k_q
exciplex; triplet states; ketones; alkenes; alkynes; quenching
61 ref.; 1 tab.

71–02 Engel, P.S.; Monroe, B.M. Complications in photosensitized reactions. Advances in Photochemistry, J.N. Pitts, Jr., G.S. Hammond, and W.A. Noyes, Jr. (eds.), John Wiley and Sons, New York, N.Y., 1971, Vol. 8, p.245–313

ϕ_F; ϕ_{isc}; k_{ET}; τ_F; τ_P; E_T; E_S
fluorescence; energy transfer; aromatics; ketones; aldehydes; carbonyl compounds; triplet states; singlet states; solid; phosphorescence
323 ref.; 12 tab.; 7 fig.

71–03 Denisov, E.T. Redox reactions of atoms and radicals with ions in solution. Russ. Chem. Rev. 40(1): 24–33 (1971)

k_{et}; k; E_a; A
hydrogen atom; radicals; metal ions; transition metal complexes; peroxy radicals; alkyl radicals; hydroxyl radical; inorganic; organic
130 ref.; 6 tab.; Data for both aqueous and nonaqueous solution.

71-04 Berlman, I.B. Handbook of fluorescence spectra of aromatic molecules. Academic Press, New York, N.Y., Second ed., 1971, 473p.

ϕ_F; τ_P; τ_F; ϵ_{abs}; ϵ_{emis}; λ_{emis}; λ_{abs}; emis. spec.; abs. spec.; *R*

heterocyclic compounds; polyphenyls; polycyclic; fluorescence; aromatics

~255 ref.; 307 spectra; Extensive bibliography containing references to 1000 works. See earlier edition [65-01].

72-01 Habersbergerova, A.; Janovsky, I.; Kourim, P. Absorption spectra of intermediates formed during radiolysis and photolysis. Part II. Radiat. Res. Rev. 4(2/3): 123-231 (1972)

ϵ_{abs}; λ_{abs}

radicals; aromatics; aliphatics; heterocyclic compounds; solvated electrons; solid; liquid; organic; inorganic

410 ref.; 14 tab.; Extensive compilation of data on radical transients (see [68-01] for part I). Data for both aqueous and nonaqueous solution.

72-02 Howard, J.A. Absolute rate constants for reactions of oxyl radicals. Advances in Free-Radical Chemistry, G.H. Williams (ed.), Academic Press, New York, N.Y., 1972, Vol. 4, p.49-173

k; k_{H-ab}; k_{com}; k_{ox}; $k_{disprop}$; E_a; *K*; *A*

radicals; hydroxyl radical; alkoxy radicals; peroxy radicals; triplet states; gas; aromatics; aldehydes; ketones; phenols, substituted; amines; metal ions; nitroxide radicals; carbonyl compounds; aryloxy radicals; aliphatics

~ 300 ref.; 50 tab.; Data for both aqueous and nonaqueous solution.

72-03 Berces, T. The decomposition of aldehydes and ketones. Chemical Kinetics, C.H. Bamford and C.F.H. Tipper (eds.), Elsevier Publishing Co., New York, N.Y., 1972, Vol. 5, p.234-380

ϕ(prod); prod. anal.

aldehydes; ketones; aliphatics; gas; excited states

333 ref.; 33 tab.; Data for both aqueous and nonaqueous solution.

73-01 Anbar, M.; Bambenek, M.; Ross, A.B. Selected specific rates of reactions of transients from water in aqueous solution. 1. Hydrated electron. NSRDS-NBS-43, May 1973, 59p.

k

aqueous; hydrated electron; organic; inorganic; metal ions; transition metal complexes; aliphatics; aromatics; biological materials

202 ref.; 4 tab Formula index.

73-02 Dorfman, L.M.; Adams, G.E. Reactivity of the hydroxyl radical in aqueous solutions. NSRDS-NBS 46, June, 1973, 59p.

k_{add}; k_{H-ab}; k_{et}

hydroxyl radical; aqueous; biological materials; oxide radical ion; radical processes; inorganic; alcohols; carboxylic acids; carbonyl compounds; organic; aliphatics; radicals

320 ref.; 21 tab.

73-03 Kirk, A.D. The photochemistry of chromium(III) complexes. Mol. Photochem. 5(2): 127-50 (1973)

ϕ_{aq}; E_a

transition metal complexes; chromium compounds; quenching; aromatics; energy transfer

61 ref.; 4 tab.; 6 fig.

73-04 Scaiano, J.C. Intermolecular photoreductions of ketones. J. Photochem. 2(2): 81-118 (1973)

ϕ_{red}; k_q; k_{H-ab}; τ_T; τ_r
gas; ketones; fluorescence; triplet states; carbonyl compounds; aliphatics; aromatics; amines; quenching; alcohols; boranes
199 ref.; 19 tab.; 3 fig.

73-05 Murov, S.L. Handbook of photochemistry. Dekker, New York, 1973, 272p.

ϕ_F; ϕ_P; ϕ_{isc}; k_q; k_{ET}; k_{H-ab}; τ_T; τ_F; τ_P; E_S; E_T; ϵ_{abs}; μ
quenching; aromatics; ketones; gas; solid: organic; energy transfer; triplet states; singlet states; carbonyl compounds; alkenes
Book of data tables covering materials of interest to photochemists. Bibliography. Indexes. Data for both aqueous and nonaqueous solution.

73-06 Birks, J.B. The spectroscopy of the π-electronic states of aromatic hydrocarbons. Organic Molecular Photophysics, J.B. Birks (ed.), John Wiley and Sons, New York, N.Y., 1973, Vol. 1, p.1-55

ϕ_P; k_{isc}; k_F; k_{nr}; τ_r; τ_T; E_S; E_T; E_a; ΔE(S-S); ΔE(T-T); ϵ_{abs}(S-S); ϵ_{abs}(T-T)
aromatics; singlet states; triplet states; fluorescence; polycyclic; singlet-singlet transitions; singlet-triplet transitions; triplet-triplet transitions
98 ref.; 16 tab.

73-07 Labhart, H.; Heinzelmann, W. Triplet-triplet absorption spectra of organic molecules. Organic Molecular Photophysics, J.B. Birks (ed.), John Wiley and Sons, New York, N.Y., 1973, Vol. 1, p.297-355

ΔE(T-T); ϵ_{abs}(T-T)
triplet-triplet transitions; carbonyl compounds; aromatics; dyes; heterocyclic compounds; porphyrins; biological materials; solid
250 ref.; 9 tab.; 5 fig.

73-08 Engel, P.S.; Steel, C. Photochemistry of aliphatic azo compounds in solution. Acc. Chem. Res. 6: 275-81 (1973)

ϕ(prod)
azo-compounds; singlet states; aliphatics; triplet states
69 ref.; 3 tab.; 3 fig.

73-09 Chapman, O.L.; Weiss, D.S. Photochemistry of cyclic ketones. Organic Photochemistry, O.L. Chapman (ed.), Marcel Dekker, Inc., New York, N.Y., 1973, Vol. 3, p.197-288

ϕ; ϕ_P; ϕ_{isom}; k_{H-ab}; τ_T; τ_F; λ_{abs}; ϵ_{abs}
ketones; alkenes; triplet states; singlet states; carbonyl compounds; solid
222 ref.; 14 tab.

73-10 Kirmse, W. Carbene and carbenoid formation and reactions. Chemical Kinetics, C.H. Bamford and C.F.H. Tipper (eds.), Elsevier Scientific Publishing Co., New York, N.Y., 1973, Vol. 9, p.373-415

ϕ(prod); k
carbenes; azo-compounds; alkanes; alkenes; gas; singlet states; liquid; triplet states; biradicals; ethers
140 ref.; 22 tab.

73-11 Berlman, I.B. Energy transfer parameters of aromatic compounds. Academic Press, New York, N.Y., 1973, 379p.

R
organics; aromatics; heterocyclic compounds; energy transfer
~ 450 ref.; Extensive tables of donor-acceptor distances for efficient energy transfer.

74-01 Hayon, E.; Simic, M. Acid-base properties of free radicals in solution. Acc. Chem. Res. 7(4): 114-21 (1974)

pK(radicals)
radicals; acidity; inorganic; organic; alkyl radicals; biological materials; hydroxyalkyl radicals; aqueous
108 ref.; 5 tab.; 1 fig.

74–02 Wrighton, M. The photochemistry of metal carbonyls. Chem. Rev. 74(4): 401–30 (1974)

ϕ_r; ϕ_{sub}; ϕ; ϕ_{isom}; τ_r; λ_{abs}; λ_{emis}; ϵ_{abs}; prod. anal.
metal carbonyls; solid; luminescence; alkenes; dienes; chromium compounds; molybdenum compounds; tungsten compounds; vanadium compounds; niobium compounds; tantalum compounds; iron compounds; ruthenium compounds; osmium compounds; rhenium compounds; transition metal complexes
273 ref.; 25 tab.; 7 fig.

74–03 Denisov, E.T. Liquid-phase reaction rate constants. IFI/Plenum, New York, N.Y., 1974, 771p.

k_{dis}; k_{com}; k_{isom}; k_{sub}; k_{hal-ab}; k_{H-ab}; k_{add}; $k_{disprop}$; E_a; A
radicals; radical processes; organic; hydrogen atom; hydroxyl radical; alkyl radicals; aromatics; aliphatics; alkoxy radicals; aryloxy radicals; peroxy radicals; nitroxide radicals; halogen atoms; alkenes
1385 ref.; 114 tab.; The three chapters of this book are entitled (1) Reactions of Molecules, (2) Reactions of Free Atoms and Radicals, and (3) Ionic Homolytic Reactions. Data for both aqueous and nonaqueous solution.

74–04 Hendry, D.G.; Mill, T.; Piszkiewicz, L.; Howard, J.A.; Eigenmann, H.K. A critical review of H-atom transfer in the liquid phase: Chlorine atom, alkyl, trichloromethyl, alkoxy, and alkylperoxy radicals. J. Phys. Chem. Ref. Data 3(4): 937–78 (1974)

k_{H-ab}; E_a; A
carbonyl compounds; radicals; liquid; alkanes; alkenes; silanes; organic; halogen atoms; alcohols; ethers; alkoxy radicals; alkyl radicals; peroxy radicals
149 ref.; 39 tab.; 2 fig.

74–05 Standard fluorescence spectra, 4 Volumes. Satdler Research Laboratories, Philadelphia, PA, 1974

λ_{emis}; emis. spec.; excit. spec.
fluorescence; aromatics
4 volumes containing spectra for 1000 compounds. Data for both aqueous and nonaqueous solution.

75–01 Ross, A.B. Selected specific rates of reactions of transients from water in aqueous solution. Hydrated electron, supplemental data. NSRDS-NBS 43, Supplement, June 1975, 43p.

k
aqueous; hydrated electron; organic; inorganic
163 ref.; 3 tab.; Formula index. This volume is a Supplement to [73–01].

75–02 Anbar, M.; Farhataziz Ross, A.B. Selected specific rates of reactions of transients from water in aqueous solution. II. Hydrogen atom. NSRDS-NBS-51, U.S. Dept. of Commerce, National Bureau of Standards, Washington, D.C., 1975, 56p.

k
aqueous; radicals; hydrogen atom; inorganic; organic
219 ref.; 4 tab.; Formula index.

75–03 Khudyakov, I.V.; Kuz'min, V.A. Short-lived phenoxy- and semiquinone radicals. Russ. Chem. Rev. (Engl. Transl.) 44(10): 801–15 (1975)

k_{com}; $k_{disprop}$; pK(radicals); ϵ_{abs}; λ_{abs}
aryloxy radicals; semiquinones; phenols, substituted; quinones; metal ions
205 ref.; 4 tab.; Data for both aqueous and nonaqueous solution.

75–04 Ermolaev, V.L.; Sveshnikova, E.; Shakhverdov, T.A. Energy transfer between organic molecules and transition metal ions. Russ. Chem. Rev. 44(1): 26–40 (1975)

ϕ_{aq}; k_q; k_{ET}; E_T; R
fluorescence; phosphorescence; energy transfer; metal ions; quenching; triplet states; singlet oxygen; transition metal complexes; polycyclic; solid; flavins; aromatics
110 ref.; 11 tab.; 2 fig.; Data for both aqueous and nonaqueous solution.

75–05 Cundall, R.B.; Ogilvie, S.McD. The photophysics of benzene in fluid media. Organic Molecular Photophysics. Volume 2., J.B. Birks (ed.), John Wiley & Sons, New York, 1975, p.33–93

ϕ_F; ϕ_F(EXM); ϕ_T; ϕ_{ic}; k_F; k_{ic}; k_{isc}; τ_F; τ_r; τ_{nr}; τ_F(EXM); λ_{abs}(S–S); ϵ_{abs}(S–S)
benzene; gas; deuterium compounds; fluorescence; singlet states; triplet states; liquid; excimer
196 ref.; 11 tab.; 15 fig.; Data for both aqueous and nonaqueous solution.

75–06 Wilkinson, F. Triplet quantum yields and singlet–triplet intersystem crossing. Organic Molecular Photophysics. Volume 2., J.B. Birks (ed.), John Wiley & Sons, New York, 1975, p.95–158

ϕ_P; ϕ_{isc}; ϕ_F; ϕ_T; ϕ_F(EXM); k_{isc}; τ_F; R
energy transfer; triplet states; aromatics; triplet–triplet transitions; quenching; solid; excimer; singlet states; carbonyl compounds; heterocyclic compounds; porphyrins; dyes; amines
252 ref.; 12 tab.; 15 fig.

75–07 Birks, J.B. Photophysics of aromatic molecules – a postscript. Organic Molecular Photophysics. Volume 2., J.B. Birks (ed.), John Wiley & Sons, New York, 1975, p.409–613

ϕ_F; ϕ_P; ϕ_T; k_r; k_{nr}; k_{isc}; k_q; τ_F; τ_P; τ_T; E_T; E_S; E_a; λ_{abs}; λ_{emis}; ϵ_{emis}; μ
aromatics; polycyclic; gas; liquid; excimer; singlet states; triplet states; fluorescence; phosphorescence; quenching; solid; alkanes; exciplex
618 ref.; 58 tab.; 20 fig.; Sequel, covering work published subsequent to [70–01].

75–08 Balzani, V.; Moggi, L.; Manfrin, M.F.; Bolletta, F.; Laurence, G.S. Quenching and sensitization processes of coordination compounds. Coord. Chem. Rev. 15(4): 321–433 (1975)

k_{ET}; k_q; E_T
iron compounds; molybdenum compounds; ruthenium compounds; rhodium compounds; platinum compounds; cobalt compounds; copper compounds; transition metal complexes; quenching; energy transfer; aromatics; chromium compounds; luminescence; zinc compounds; solid; manganese compounds; nickel compounds; rare–earth metal compounds; silver compounds; vanadium compounds; mercury compounds; iridium compounds; palladium compounds; excited states
216 ref.; 7 tab.; 8 fig.; Extensive catalog of excited states of coordination compounds including their modes of relaxation to the ground state. Data for both aqueous and nonaqueous solution.

76–01 Ermolaev, V.L.; Sveshnikova, E.; Shakhverdov, T.A. Investigation of the formation of complexes of organic molecules and lanthanide ions in solutions by the electronic energy transfer method. Russ. Chem. Rev. 45(10): 896–912 (1976)

k_q; k_{ET}; τ_F; R
energy transfer; luminescence; rare–earth metal ions; quenching; aromatics; fluorescence; dyes; ketones; triplet states
98 ref.; 13 tab.; 9 fig.; Data for both aqueous and nonaqueous solution.

76–02 Shlyapintokh, V.Ya.; Ivanov, V.B. The quenching of singlet oxygen. Russ. Chem. Rev. 45(2): 99–110 (1976)

k_d; k_q; k_{ox}
quenching; singlet oxygen; alkenes; aromatics; amines; transition metal complexes; gas; energy transfer
77 ref.; 11 tab.; 3 fig.; Data for both aqueous and nonaqueous solution.

76–03 Ireland, J.F.; Wyatt, P.A.H. Acid–base properties of electronically excited states of organic molecules. Advances in Physical Organic Chemistry, V. Gold and D. Bethell (eds.), Academic Press, New York, N.Y., 1976, Vol. 12, p.131–221

k_{dis}; k_{add}; pK(S); pK(T)
singlet states; triplet states; acidity; aromatics; phenols, substituted; organic; aqueous; heterocyclic compounds; carbonyl compounds; proton transfer
~260 ref.; 17 tab.; 13 fig.

76–04 Bellus, D. Quenchers of singlet oxygen. A critical review. Singlet Oxygen, Reactions with Organic Compounds and Polymers, B. Ranby and J.F. Rabek (eds.), John Wiley & Sons, New York, N.Y., 1976, p.61–110

k_q; k_{ox}; k_d
singlet oxygen; biological materials; energy transfer; amines; quenching; polyenes; phenols, substituted; organic; inorganic; transition metal complexes
231 ref.; 10 tab.; 5 fig.; Data for both aqueous and nonaqueous solution.

76–05 Gollnick, K. Mechanism and kinetics of chemical reactions of singlet oxygen with organic compounds. Singlet Oxygen, Reactions with Organic Compounds and Polymers, B. Ranby and J.F. Rabek (eds.), John Wiley & Sons, New York, N.Y., 1976, p.111–34

k_q; k_{ox}; τ_d
singlet oxygen; gas; organic; amines; dienes; aromatics; heterocyclic compounds; biological materials; alkenes
150 ref.; 10 tab.

77–01 Farhataziz; Ross, A.B. Selected specific rates of reactions of transients from water in aqueous solution. III. Hydroxyl radical and perhydroxyl radical and their radical ions. NSRDS–NBS–59, Jan. 1977, 122p.

k
aqueous; hydroxyl radical; perhydroxyl radical; radicals; oxide radical ion; superoxide ion; inorganic; organic
~ 460 ref.; 6 tab.; Formula index.

77–02 Martynov, I.Yu.; Demyashkevich, A.B.; Uzhinov, B.M.; Kuz'min, M.G. Proton transfer reactions in the excited electronic states of aromatic molecules. Russ. Chem. Rev. 46(1): 1–15 (1977)

ϕ; k_{add}; k_q; pK(S); pK(T)
singlet states; triplet states; aromatics; phenols, substituted; amines; proton transfer; heterocyclic compounds; acidity; carboxylic acids
186 ref.; 6 tab.; 3 fig.; Data for both aqueous and nonaqueous solution.

77–03 von Sonntag, C.; Schuchmann, H.P. Photolysis of saturated alcohols, ethers, and amines. Advances in Photochemistry, J.N. Pitts, Jr., G.S. Hammond and K. Gollnick (eds.), Wiley, New York, 1977, Vol. 10, p.59–145.

ϕ(prod); ϵ_{abs}; prod. anal.
alcohols; ethers; amines; liquid; aqueous; gas; solid; aliphatics
295 ref.; 27 tab.; Data for both aqueous and nonaqueous solution.

77–04 Cundall, R.B.; Robinson, D.A.; Pereira, L.C. Excitation and deexcitation of benzene. Advances in Photochemistry, J.N. Pitts, Jr., G.S. Hammond and K. Gollnick (eds.), Wiley, New York, 1977, Vol. 10, p.147–219

ϕ_F; ϕ_T; ϕ_F(EXM); k_F; τ_F; τ_P; τ(EXM); abs. spec.; emis. spec.; I
benzene; singlet states; triplet states; gas; fluorescence; excimer; solid
288 ref.; 15 tab.; 10 fig.

77-05 Bock, C.R.; Koerner von Gustorf, E.A. Primary photoprocesses of organo-transition metal compounds. Advances in Photochemistry, J.N. Pitts, Jr., G.S. Hammond and K. Gollnick (eds.), Wiley, New York, 1977, Vol. 10, p.221-310

ϕ_{sub}; ϕ_{isom}; ϕ_{add}; ϕ(prod); ϕ_r; k_{add}; k_{dis}; k_q; k_r; k_{nr}; τ_r; λ_{emis}
triplet states; transition metal complexes; solid; metal carbonyls; luminescence; quenching; metal-metal bonded complexes
262 ref.; 13 tab.; 12 fig.

77-06 De Schryver, F.C.; Boens, N.; Put, J. Excited state behavior of some bichromophoric systems. Advances in Photochemistry, J.N. Pitts, Jr., G.S. Hammond and K. Gollnick (eds.), Wiley, New York, 1977, Vol. 10, p.359-465

ϕ_F; ϕ_{isom}; ϕ_{isc}; k_{isom}; k_{ET}; λ_{emis}
singlet states; triplet states; excimer; alkenes; fluorescence; energy transfer; ketones; polymerization; aromatics
224 ref.; 24 tab ; 13 fig.

78-01 Buxton, G.V.; Sellers, R.M. Compilation of rate constants for the reactions of metal ions in unusual valency states. NSRDS-NBS 62, 1978, 78p.

k
aqueous; transition metal complexes; metal ions; radicals
96 ref.; 25 tab.

78-02 Maciejewski, A.; Matuszewski, B. Intersystem crossing quantum yield ϕ_{isc} of electronically excited molecules. Part II. Effect of some physicochemical factors on the values of ϕ_{isc} and comparison of organic compounds for which these values were determined. Wiad. Chem. 32(3): 151-65 (1978)

ϕ_{isc}
singlet states; triplet states
146 ref.; 1 table listing 200 compounds with references in which data may be found.

78-03 Swallow, A.J. Reactions of free radicals produced from organic compounds in aqueous solution by means of radiation. Prog. React. Kinet. 9(3/4): 195-366 (1978)

k; k_{com}; k_{add}; k_{et}; k_{elim}; pK(radicals); E_a; E°
aqueous; radicals; radical processes; organic; alkyl radicals; metal ions; proton transfer; hydroxyalkyl radicals; inorganic; semiquinones; biological materials
360 ref.; 58 tab.

78-04 Scaiano, J.C.; Lissi, E.A.; Encina, M.V. Chemistry of the biradicals produced in the Norrish type II reaction. Rev. Chem. Intermediates 2: 139-95 (1978)

ϕ_{dis}; ϕ_{isom}; k_{H-ab}; k_{et}; k; k_{add}; τ_T; τ(B); E_a; A
biradicals; Norrish type II processes; ketones; triplet states; singlet states; gas; aliphatics; aromatics
288 ref.; 20 tab.; 3 fig.

78-05 Balzani, V.; Bolletta, F.; Gandolfi, M.T. Maestri, M. Bimolecular electron transfer reactions of the excited states of transition metal complexes. Top. Curr. Chem. 75: 1-64 (1978)

k_q; τ_r; E°; ΔE; λ_{abs}
quenching; chromium compounds; ruthenium compounds; transition metal complexes; osmium compounds; iridium compounds; aromatics; uranium compounds
314 ref.; 6 tab.; 22 fig.

78-06 Bensasson, R.; Land, E.J. Physical properties of excited states: A general method for measuring triplet-triplet extinction coefficients, singlet-triplet intersystem crossing efficiencies, and related parameters. Photochemical and Photobiological Reviews, K.C. Smith(ed.), Plenum Press, New York, N.Y., 1978, Vol 3, p.163-91

ϕ_{isc}; λ_{abs}(S-S); λ_{abs}(T-T); ϵ_{abs}(S-S); ϵ_{abs}(T-T)
triplet states; singlet states; aromatics; polycyclic; biological materials
68 ref.; 4 tab.; Data for both aqueous and nonaqueous solution.

79-01 Ross, A.B.; Neta, P. Rate constants for reactions of inorganic radicals in aqueous solution. NSRDS-NBS 65, 1979, 55p.

k

aqueous; radicals; inorganic; organic; aliphatics; aromatics

95 ref.; 9 tab.; Data for over 560 radical reactions. Formula index.

79-02 Fox, M.A. The photoexcited states of organic anions. Chem. Rev. 79(3): 253-73 (1979)

λ_{abs}

carbanions; radicals; alkali metal ions

279 ref.; 7 tab.; 3 fig.; Data for both aqueous and nonaqueous solution.

79-03 Bellus, D. Physical quenchers of singlet molecular oxygen. Advances in Photochemistry, J.N. Pitts, Jr., G.S. Hammond and K. Gollnick (eds.), John Wiley and Sons, New York, N.Y., 1979, Vol. 11, p.105-205

k; k_q; k_d; τ_d; E_T

singlet oxygen; organic; quenching; alkenes; amines; biological materials; phenols, substituted; transition metal complexes; heterocyclic compounds

332 ref.; 11 tab.; 7 fig.; Data for both aqueous and nonaqueous solution.

79-04 Gollnick, K.; Kuhn, H.J. Ene-reactions with singlet oxygen. Singlet Oxygen, H.H. Wasserman and R.W. Murray (eds.), Academic Press, New York, N.Y., 1979, Vol. 40 of Organic Chemistry, p. 287-427

k; τ_d; E_a; prod. anal.; I.P.

alkenes; singlet oxygen; gas

286 ref.; 15 tab.; 4 fig.; Includes an extensive catalog of product yields for the reaction of singlet oxygen with various alkenes.

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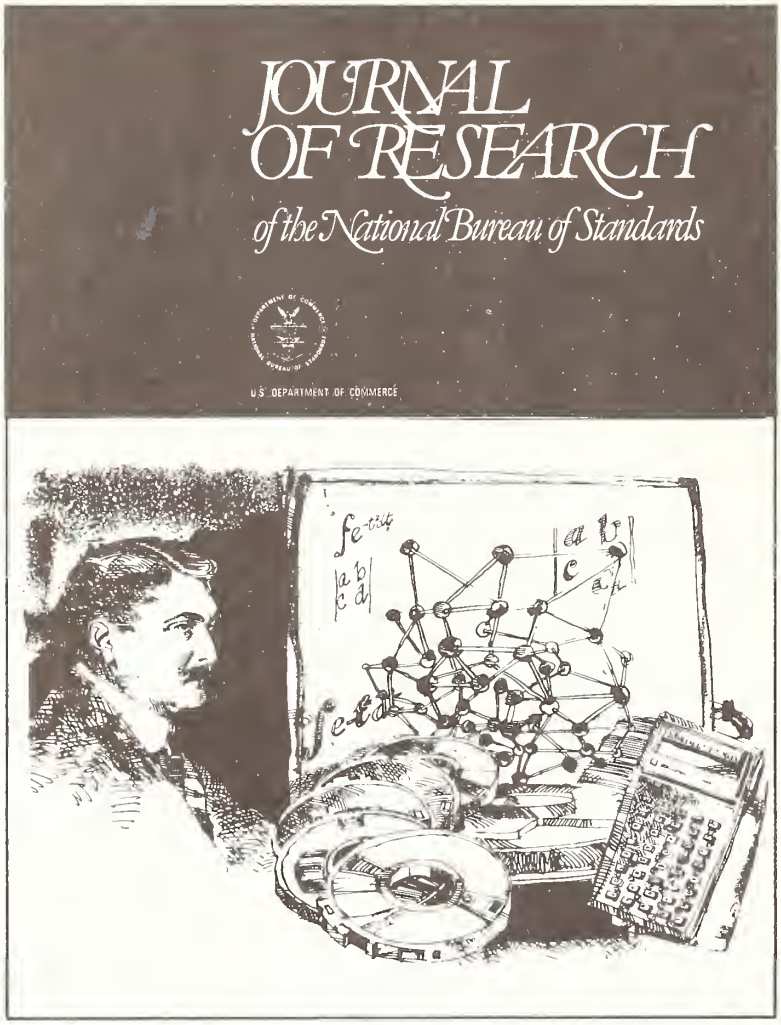

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